



GEORGIA SECTION

# MEETING Announcement



**Tuesday, October 28<sup>th</sup> 2014**

**Dr. John Bacsa**

A brief historical tour of chemical X-ray crystallography: from Max von Laue's discovery of diffraction of X-rays to modern charge density analysis.

**Location:**

Maggiano's Little Italy– Buckhead  
3368 Peachtree Rd, Atlanta, GA 30326

**Directions:** [Click Here](#)

**Tuesday, October 28<sup>th</sup> 2014**

**6:00 pm** Meet and mingle (cash bar)

**6:20 pm** Dinner

**7:30 pm** Presentation

**Menu (Family Style):**

**First Course:**

Maggiano's Salad & Caesar Salad  
Bruschetta & Calamari

**Second Course:**

Spaghetti Marinara & Four cheese Ravioli  
Salmon & Chicken Parmesan

**Desserts:**

Cheesecake & Profiteroles

\*\*Coffee, Soda & Ice Tea included

**RSVP by 4:00 pm on 24 October 2014**  
to Joel Pollino at [joelpollino@gmail.com](mailto:joelpollino@gmail.com)

**Price:**

\$30 regular; \$15 students, K-12 teachers, retired  
current ACS members

**Payment: At the door**

Cash, Credit Card, or Check to: "Georgia Section  
ACS"



**Dr. John Bacsa**

**Facilities Director, X-Ray Diffraction Center  
Emory University, Atlanta GA**



**Dr. John Bacsa** is the Facilities Director, X-ray Diffraction Center at Emory University. The X-ray center's primary function is the complete determination of the three-dimensional arrangement of atoms and molecules in inorganic, organometallic, organic and biological compounds. Laboratories at Emory and elsewhere rely on our service for the unambiguous structural characterization of their products, to establish relative and absolute stereochemistry and to confirm the outcomes of reactions. He holds a PhD from the University of the Witwatersrand, Johannesburg, South Africa (advisor: Prof. J.C.A. Boeyens) where he studied the



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effects of steric interactions on rhenium-rhenium bond distances by combination of X-ray crystallography and molecular modelling. With these results he was able to determine the relationships between bond strengths, harmonic force-constants and characteristic bond lengths for these metal-metal bonds. He is interested in determining and analyzing the electron density in molecules from theory and experiment and in determining bond order functions for varied elements in the periodic table. He has approximately 133 publications that have been cited 722 times.

## Positions and Honors.

### Positions:

**1998-01** Managed the X-ray facility, University of Cape Town, South Africa.

**2002-02** Research associate of Prof. Mark D. Hollingsworth, Kansas State University.

**2002-04** Postdoctoral Research in Inorganic Chemistry, Texas A&M University.

**2004-2007** Manager, Toronto Protein Diffraction Facilities, Sickkids Hospital, Toronto, Canada.

**2007-2011** Senior Experimental Officer, University of Liverpool, UK.

**2004-present** Facilities director of the X-ray Diffraction Center at Emory University.

### Awards:

**2002** Fellowship from National Aeronautics and Space Administration (NASA).

**2002-2004** Reviewer, *Inorganic Chemistry*.

**2002-present** Reviewer, *Acta Crystallographica* Section C.

**2012** Research highlighted in *Chemical Engineering News*.

## A brief historical tour of chemical X-ray crystallography: from Max von Laue's discovery of diffraction of X-rays to modern charge density analysis

**Abstract:** The aim of this presentation is to give, from a historical perspective, a brief overview of X-ray crystallography. Although Bragg is credited for determining the first atomic resolution structure of a crystal by X-rays, Max von Laue demonstrated that crystals diffract X-rays and was awarded the Nobel Prize for Physics in 1914 for this discovery. Von Laue's X-ray diffraction photos revealed the underlying space-group symmetry of crystals and validated mathematical derivations of the space lattice. Subsequently, Bragg made the connection between wavelength of X-ray light and the spacing between atoms in a crystal, and opened the door to routine structure determinations. The increase in the data quality allowed the nature of chemical interactions to be fundamentally characterized. Charge and energy can now be precisely determined by X-ray crystallography but the exact nature of a chemical bond remains elusive.